

Acta Crystallographica Section E

## Structure Reports

Online

ISSN 1600-5368

## 2-Benzylsulfanyl-4-[(4-methylphenyl)-sulfanyl]-6-pentylpyrimidine-5-carbonitrile

Ali A. El-Emam,<sup>a,‡</sup> Omar A. Al-Deeb,<sup>a</sup> Nasser R. El-Brollosy,<sup>a</sup> Seik Weng Ng<sup>b,c</sup> and Edward R. T. Tiekink<sup>b\*</sup>

<sup>a</sup>Department of Pharmaceutical Chemistry, College of Pharmacy, King Saud University, Riyadh 11451, Saudi Arabia, <sup>b</sup>Department of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia, and <sup>c</sup>Chemistry Department, Faculty of Science, King Abdulaziz University, PO Box 80203 Jeddah, Saudi Arabia  
Correspondence e-mail: Edward.Tiekink@gmail.com

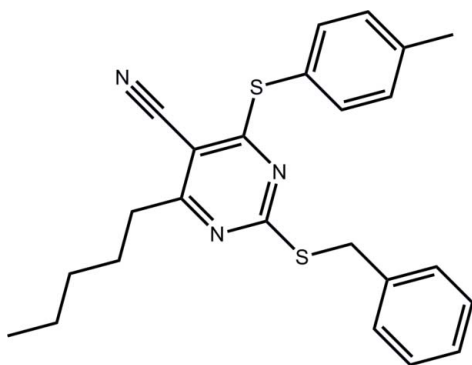
Received 1 June 2012; accepted 3 June 2012

Key indicators: single-crystal X-ray study;  $T = 294$  K; mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å;  $R$  factor = 0.044;  $wR$  factor = 0.138; data-to-parameter ratio = 18.0.

In the title compound,  $\text{C}_{24}\text{H}_{25}\text{N}_3\text{S}_2$ , the S-bound benzene rings have orthogonal [dihedral angle =  $85.31(9)^\circ$ ] and splayed [ $67.92(11)^\circ$ ] orientations with respect to the pyrimidine ring; the dihedral angle between the benzene rings is  $48.18(12)^\circ$ . The pentyl group has an extended all-*trans* conformation and lies to one side of the pyrimidine ring [the  $\text{N}_{\text{py}}-\text{C}_{\text{py}}-\text{C}_{\text{p}}-\text{C}_{\text{p}}$  torsion angle =  $-85.7(2)^\circ$ ; py = pyrimidine and p = pentyl].

### Related literature

For the chemotherapeutic activity of pyrimidine derivatives see: Ghoshal & Jacob (1997); De Corte (2005); Al-Omar *et al.* (2010); Al-Abdullah *et al.* (2011); Al-Turkistani *et al.* (2011). For a related pyrimidine structure, see: El-Emam *et al.* (2012).



<sup>‡</sup> Additional correspondence author, e-mail: elemam5@hotmail.com.

### Experimental

#### Crystal data

$\text{C}_{24}\text{H}_{25}\text{N}_3\text{S}_2$   
 $M_r = 419.59$   
 Monoclinic,  $P2_1/c$   
 $a = 9.9178(2)$  Å  
 $b = 8.2235(2)$  Å  
 $c = 28.4388(8)$  Å  
 $\beta = 96.115(2)^\circ$   
 $V = 2306.24(10)$  Å<sup>3</sup>  
 $Z = 4$   
 Cu  $K\alpha$  radiation  
 $\mu = 2.19$  mm<sup>-1</sup>  
 $T = 294$  K  
 $0.40 \times 0.20 \times 0.10$  mm

#### Data collection

Agilent SuperNova Dual diffractometer with Atlas detector  
 Absorption correction: multi-scan (*CrysAlis PRO*; Agilent, 2012)  
 $T_{\text{min}} = 0.689$ ,  $T_{\text{max}} = 1.000$   
 9523 measured reflections  
 4746 independent reflections  
 3720 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.017$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.044$   
 $wR(F^2) = 0.138$   
 $S = 1.03$   
 4746 reflections  
 264 parameters  
 H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.18$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.24$  e Å<sup>-3</sup>

Data collection: *CrysAlis PRO* (Agilent, 2012); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *DIAMOND* (Brandenburg, 2006); software used to prepare material for publication: *publCIF* (Westrip, 2010).

The financial support of the Deanship of Scientific Research and the Research Center of the College of Pharmacy, King Saud University, is greatly appreciated. We also thank the Ministry of Higher Education (Malaysia) for funding structural studies through the High-Impact Research scheme (UM-C/HIR/MOHE/SC/12).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: ZL2483).

### References

- Agilent (2012). *CrysAlis PRO*. Agilent Technologies, Yarnton, England.  
 Al-Abdullah, E. S., Al-Obaid, A. M., Al-Deeb, O. A., Habib, E. E. & El-Emam, A. A. (2011). *Eur. J. Med. Chem.* **46**, 4642–4647.  
 Al-Omar, M. A., Al-Obaid, A. M., El-Brollosy, N. R. & El-Emam, A. A. (2010). *Synth. Commun.* **40**, 1530–1538.  
 Al-Turkistani, A. A., Al-Deeb, O. A., El-Brollosy, N. R. & El-Emam, A. A. (2011). *Molecules*, **16**, 4764–4774.  
 Brandenburg, K. (2006). *DIAMOND*. Crystal Impact GbR, Bonn, Germany.  
 De Corte, B. L. (2005). *J. Med. Chem.* **48**, 1689–1696.  
 El-Emam, A. A., Al-Deeb, O. A., Al-Turkistani, A. A., Ng, S. W. & Tiekink, E. R. T. (2012). *Acta Cryst.* **E68**, o2055–o2056.  
 Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.  
 Ghoshal, K. & Jacob, S. T. (1997). *Biochem. Pharmacol.* **53**, 1569–1575.  
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.  
 Westrip, S. P. (2010). *J. Appl. Cryst.* **43**, 920–925.

## supplementary materials

*Acta Cryst.* (2012). E68, o2054 [doi:10.1107/S1600536812025275]

**2-Benzylsulfanyl-4-[(4-methylphenyl)sulfanyl]-6-pentylpyrimidine-5-carbonitrile**

Ali A. El-Emam, Omar A. Al-Deeb, Nasser R. El-Brollosy, Seik Weng Ng and Edward R. T. Tiekink

**Comment**

The ability of pyrimidine derivatives to inhibit vital enzymes responsible for DNA bio-synthesis is the reason behind their chemotherapeutic efficacy. Thus, several pyrimidine non-nucleoside derivatives exhibit anti-cancer (Ghoshal & Jacob, 1997), anti-viral (De Corte, 2005) and anti-bacterial activities (Al-Abdullah *et al.*, 2011). The synthesis and crystal structure determination of the title compound, was undertaken in connection with on-going studies of the chemical, pharmacological and structural properties of pyrimidine derivatives (Al-Omar *et al.*, 2010; Al-Turkistani *et al.*, 2011; El-Emam *et al.* 2012).

With reference to the pyrimidine ring (r.m.s. deviation = 0.011 Å) in the title compound, Fig. 1, the S1- and S2-bound benzene rings form dihedral angles of 85.31 (9) and 67.92 (11)°, respectively, which indicate an orthogonal and a splayed orientation, respectively; the dihedral angle between the benzene rings is 48.18 (12)°. The pentyl group lies to one side of the pyrimidine ring with the N2—C3—C20—C21 torsion angle being -85.7 (2)°. The remaining chain has an extended all-*trans* conformation [the C20—C21—C22—C23 and C21—C22—C23—C24 torsion angles are -173.8 (2) and 179.6 (2)°, respectively].

No specific intermolecular interactions are evident in the crystal structure. Globally, molecules lie in layers in the *ab* plane which stack along the *c* axis, Fig. 2.

**Experimental**

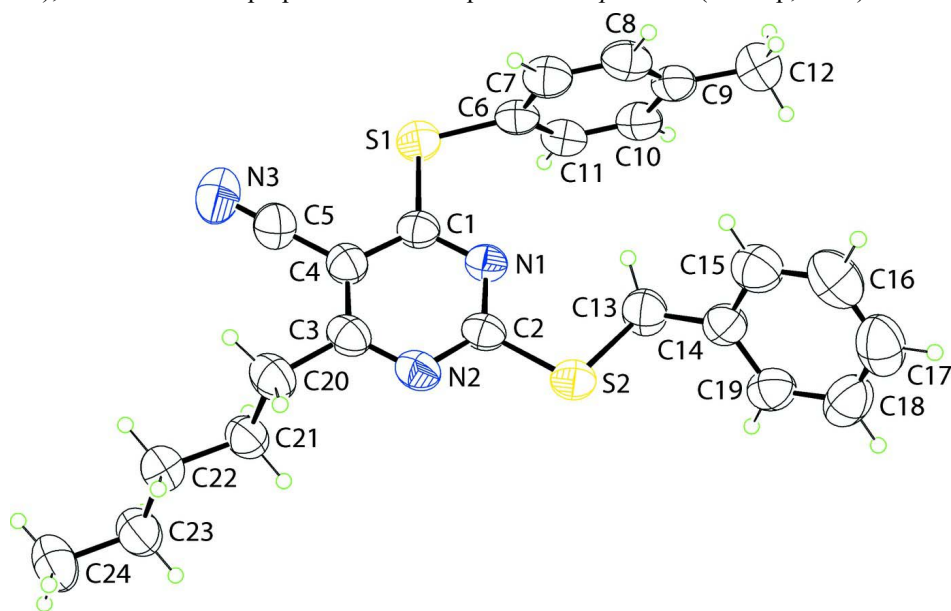
To a solution of 2-(benzylsulfanyl)-4-chloro-6-(*n*-pentyl)pyrimidine-5-carbonitrile (665 mg, 2 mmol) in dry pyridine (3 ml), 4-thiocresol (248 mg, 2 mmol) was added and the mixture was heated under reflux for 6 h. On cooling, the solvent was then distilled off *in vacuo* and water (5 ml) was added to the residue. The separated precipitate was collected by filtration, washed with cold water, dried and recrystallized from ethanol to yield 747 mg (89%) of the title compound as colourless crystals. M.p.: 384–386 K. Single crystals suitable for X-ray analysis were obtained by slow evaporation of a solution of the title compound in CHCl<sub>3</sub>:EtOH (1:1, 5 ml) at room temperature. <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>, 500.13 MHz): δ 0.85 (t, 3H, CH<sub>3</sub>, J = 7.0 Hz), 1.26–1.34 (m, 4H, CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 1.64–1.69 (m, 2H, CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 2.27 (s, 3H, Ar—CH<sub>3</sub>), 2.76 (t, 2H, CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>, J = 7.0 Hz), 4.02 (s, 2H, CH<sub>2</sub>S), 6.98–6.99 (m, 2H, Ar—H), 7.21–7.23 (m, 3H, Ar—H), 7.28 (d, 2H, Ar—H, J = 8.0 Hz), 7.51 (d, 2H, Ar—H, J = 8.0 Hz). <sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>, 125.76 MHz): δ 14.16 (CH<sub>3</sub>), 21.29 (CH<sub>2</sub>CH<sub>3</sub>), 22.29 (ArCH<sub>3</sub>), 27.39 (CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 31.13 (CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 34.51 (CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 36.12 (CH<sub>2</sub>S), 99.14 (C-5), 114.69 (CN), 122.19, 127.56, 128.74, 129.01, 130.67, 136.20, 137.58, 140.99 (Ar—C), 172.81, 173.15, 173.37 (C-2, C-4 & C-6).

**Refinement**

Carbon-bound H-atoms were placed in calculated positions [C—H = 0.93 to 0.97 Å, *U*<sub>iso</sub>(H) = 1.2–1.5*U*<sub>eq</sub>(C)] and were included in the refinement in the riding model approximation.

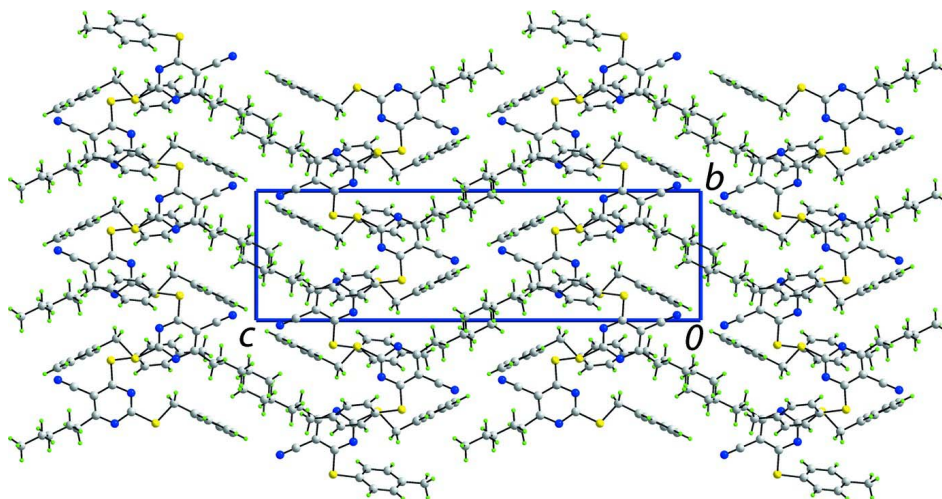
**Computing details**

Data collection: *CrysAlis PRO* (Agilent, 2012); cell refinement: *CrysAlis PRO* (Agilent, 2012); data reduction: *CrysAlis PRO* (Agilent, 2012); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *DIAMOND* (Brandenburg, 2006); software used to prepare material for publication: *publCIF* (Westrip, 2010).



**Figure 1**

The molecular structure of the title compound showing the atom-labelling scheme and displacement ellipsoids at the 35% probability level.



**Figure 2**

A view in projection down the *a* axis of the unit-cell contents for the title compound.

## 2-Benzylsulfanyl-4-[(4-methylphenyl)sulfanyl]-6-pentylpyrimidine-5-carbonitrile

## Crystal data

$C_{24}H_{25}N_3S_2$	$F(000) = 888$
$M_r = 419.59$	$D_x = 1.208 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/c$	Cu $K\alpha$ radiation, $\lambda = 1.54184 \text{ \AA}$
Hall symbol: -P 2ybc	Cell parameters from 3374 reflections
$a = 9.9178 (2) \text{ \AA}$	$\theta = 4.6\text{--}76.5^\circ$
$b = 8.2235 (2) \text{ \AA}$	$\mu = 2.19 \text{ mm}^{-1}$
$c = 28.4388 (8) \text{ \AA}$	$T = 294 \text{ K}$
$\beta = 96.115 (2)^\circ$	Prism, colourless
$V = 2306.24 (10) \text{ \AA}^3$	$0.40 \times 0.20 \times 0.10 \text{ mm}$
$Z = 4$	

## Data collection

Agilent SuperNova Dual	$T_{\min} = 0.689, T_{\max} = 1.000$
diffractometer with Atlas detector	9523 measured reflections
Radiation source: SuperNova (Cu) X-ray	4746 independent reflections
Source	3720 reflections with $I > 2\sigma(I)$
Mirror monochromator	$R_{\text{int}} = 0.017$
Detector resolution: $10.4041 \text{ pixels mm}^{-1}$	$\theta_{\max} = 76.7^\circ, \theta_{\min} = 5.6^\circ$
$\omega$ scan	$h = -12 \rightarrow 10$
Absorption correction: multi-scan	$k = -10 \rightarrow 10$
( <i>CrysAlis PRO</i> ; Agilent, 2012)	$l = -35 \rightarrow 29$

## Refinement

Refinement on $F^2$	Hydrogen site location: inferred from neighbouring sites
Least-squares matrix: full	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.044$	$w = 1/[\sigma^2(F_o^2) + (0.0721P)^2 + 0.1537P]$
$wR(F^2) = 0.138$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.03$	$(\Delta/\sigma)_{\max} = 0.001$
4746 reflections	$\Delta\rho_{\max} = 0.18 \text{ e \AA}^{-3}$
264 parameters	$\Delta\rho_{\min} = -0.24 \text{ e \AA}^{-3}$
0 restraints	Extinction correction: <i>SHELXL</i> ,
Primary atom site location: structure-invariant direct methods	$Fc^* = kFc[1 + 0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$
Secondary atom site location: difference Fourier map	Extinction coefficient: 0.0040 (4)

## Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.78263 (6)	0.31009 (7)	0.67270 (2)	0.0889 (2)
S2	0.60772 (5)	0.79595 (6)	0.773134 (19)	0.08036 (18)

---

N1	0.68705 (13)	0.56138 (17)	0.71826 (5)	0.0665 (3)
N2	0.58228 (15)	0.79774 (19)	0.68118 (6)	0.0764 (4)
N3	0.6892 (3)	0.4609 (4)	0.55333 (8)	0.1323 (9)
C1	0.70156 (16)	0.4995 (2)	0.67609 (7)	0.0696 (4)
C2	0.62755 (16)	0.7066 (2)	0.71888 (7)	0.0678 (4)
C3	0.59644 (18)	0.7347 (3)	0.63894 (8)	0.0775 (5)
C4	0.65498 (17)	0.5814 (3)	0.63450 (7)	0.0745 (5)
C5	0.6733 (2)	0.5142 (3)	0.58931 (8)	0.0940 (6)
C6	0.84843 (18)	0.2845 (2)	0.73261 (7)	0.0729 (5)
C7	0.97186 (19)	0.3532 (3)	0.74887 (8)	0.0858 (6)
H7	1.0217	0.4096	0.7283	0.103*
C8	1.0201 (2)	0.3376 (3)	0.79569 (9)	0.0886 (6)
H8	1.1030	0.3847	0.8064	0.106*
C9	0.9493 (2)	0.2538 (3)	0.82738 (8)	0.0825 (5)
C10	0.8272 (2)	0.1824 (3)	0.80980 (8)	0.0846 (6)
H10	0.7786	0.1229	0.8301	0.102*
C11	0.77708 (18)	0.1978 (2)	0.76337 (8)	0.0787 (5)
H11	0.6948	0.1497	0.7525	0.094*
C12	1.0013 (3)	0.2400 (4)	0.87868 (10)	0.1206 (9)
H12A	1.0233	0.1287	0.8861	0.181*
H12B	0.9329	0.2767	0.8977	0.181*
H12C	1.0811	0.3059	0.8851	0.181*
C13	0.6781 (3)	0.6417 (3)	0.81484 (8)	0.0932 (6)
H13A	0.6102	0.5595	0.8189	0.112*
H13B	0.7550	0.5892	0.8028	0.112*
C14	0.7221 (2)	0.7210 (2)	0.86107 (7)	0.0781 (5)
C15	0.8585 (3)	0.7470 (4)	0.87499 (10)	0.1021 (7)
H15	0.9228	0.7152	0.8553	0.123*
C16	0.8998 (3)	0.8197 (4)	0.91789 (13)	0.1233 (10)
H16	0.9918	0.8365	0.9267	0.148*
C17	0.8086 (4)	0.8667 (4)	0.94725 (11)	0.1249 (10)
H17	0.8378	0.9148	0.9761	0.150*
C18	0.6744 (4)	0.8435 (3)	0.93449 (10)	0.1126 (8)
H18	0.6112	0.8762	0.9545	0.135*
C19	0.6317 (2)	0.7711 (3)	0.89167 (9)	0.0911 (6)
H19	0.5394	0.7557	0.8833	0.109*
C20	0.5482 (2)	0.8344 (3)	0.59621 (8)	0.0928 (6)
H20A	0.5996	0.8045	0.5704	0.111*
H20B	0.5653	0.9484	0.6033	0.111*
C21	0.3988 (2)	0.8112 (3)	0.58083 (8)	0.0843 (5)
H21A	0.3462	0.8583	0.6043	0.101*
H21B	0.3785	0.6959	0.5787	0.101*
C22	0.3585 (2)	0.8912 (3)	0.53285 (7)	0.0885 (6)
H22A	0.3891	1.0033	0.5342	0.106*
H22B	0.4049	0.8360	0.5091	0.106*
C23	0.2100 (2)	0.8882 (4)	0.51799 (9)	0.1024 (7)
H23A	0.1632	0.9429	0.5417	0.123*
H23B	0.1792	0.7762	0.5162	0.123*
C24	0.1729 (3)	0.9695 (4)	0.47047 (8)	0.1191 (9)

H24A	0.0762	0.9674	0.4629	0.179*
H24B	0.2151	0.9124	0.4465	0.179*
H24C	0.2038	1.0802	0.4719	0.179*

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
S1	0.0998 (4)	0.0797 (3)	0.0874 (4)	0.0206 (3)	0.0105 (3)	-0.0054 (3)
S2	0.0877 (3)	0.0652 (3)	0.0881 (3)	0.0125 (2)	0.0093 (2)	0.0008 (2)
N1	0.0611 (7)	0.0595 (7)	0.0790 (9)	0.0001 (6)	0.0086 (6)	0.0046 (7)
N2	0.0701 (8)	0.0691 (9)	0.0894 (11)	0.0054 (7)	0.0059 (7)	0.0147 (8)
N3	0.160 (2)	0.153 (2)	0.0858 (13)	0.0146 (18)	0.0225 (14)	-0.0026 (15)
C1	0.0595 (8)	0.0677 (9)	0.0822 (11)	-0.0005 (7)	0.0107 (8)	0.0044 (9)
C2	0.0556 (8)	0.0622 (9)	0.0855 (11)	-0.0011 (7)	0.0069 (7)	0.0079 (8)
C3	0.0629 (9)	0.0820 (12)	0.0871 (13)	-0.0012 (9)	0.0061 (8)	0.0159 (10)
C4	0.0656 (9)	0.0816 (11)	0.0763 (11)	-0.0025 (8)	0.0076 (8)	0.0072 (9)
C5	0.0948 (14)	0.1054 (17)	0.0820 (13)	0.0045 (12)	0.0112 (11)	0.0084 (13)
C6	0.0685 (9)	0.0600 (9)	0.0909 (12)	0.0119 (8)	0.0116 (8)	0.0009 (9)
C7	0.0723 (11)	0.0789 (12)	0.1077 (16)	-0.0043 (9)	0.0165 (11)	0.0062 (11)
C8	0.0663 (10)	0.0827 (12)	0.1157 (17)	-0.0002 (9)	0.0046 (10)	-0.0039 (12)
C9	0.0767 (11)	0.0776 (11)	0.0935 (14)	0.0201 (10)	0.0096 (10)	-0.0055 (10)
C10	0.0753 (11)	0.0795 (12)	0.1015 (15)	0.0072 (9)	0.0207 (10)	0.0103 (11)
C11	0.0645 (9)	0.0689 (10)	0.1023 (15)	0.0031 (8)	0.0074 (9)	0.0035 (10)
C12	0.1118 (19)	0.148 (3)	0.0993 (18)	0.0205 (18)	-0.0022 (14)	-0.0077 (18)
C13	0.1218 (17)	0.0681 (11)	0.0882 (14)	0.0101 (11)	0.0048 (12)	0.0053 (10)
C14	0.0911 (12)	0.0633 (10)	0.0812 (12)	0.0077 (9)	0.0148 (10)	0.0089 (9)
C15	0.0923 (14)	0.1062 (17)	0.1091 (18)	0.0028 (13)	0.0160 (13)	0.0138 (15)
C16	0.114 (2)	0.117 (2)	0.132 (3)	-0.0128 (17)	-0.0209 (19)	0.0136 (19)
C17	0.173 (3)	0.0932 (18)	0.102 (2)	-0.002 (2)	-0.014 (2)	0.0037 (15)
C18	0.160 (3)	0.0864 (15)	0.0962 (17)	0.0159 (17)	0.0345 (17)	-0.0002 (13)
C19	0.0950 (14)	0.0792 (12)	0.1008 (16)	0.0110 (11)	0.0191 (12)	0.0042 (12)
C20	0.0897 (13)	0.0930 (14)	0.0945 (15)	0.0018 (11)	0.0043 (11)	0.0267 (12)
C21	0.0856 (12)	0.0820 (12)	0.0841 (13)	0.0078 (10)	0.0038 (10)	0.0106 (10)
C22	0.0912 (13)	0.0908 (14)	0.0825 (13)	0.0047 (11)	0.0050 (10)	0.0070 (11)
C23	0.0898 (13)	0.1191 (19)	0.0964 (15)	0.0044 (13)	0.0022 (11)	0.0204 (14)
C24	0.1070 (16)	0.148 (2)	0.0972 (17)	-0.0055 (17)	-0.0142 (13)	0.0290 (17)

*Geometric parameters (Å, °)*

S1—C1	1.7601 (19)	C13—H13A	0.9700
S1—C6	1.771 (2)	C13—H13B	0.9700
S2—C2	1.739 (2)	C14—C19	1.377 (3)
S2—C13	1.824 (2)	C14—C15	1.385 (3)
N1—C1	1.325 (2)	C15—C16	1.381 (4)
N1—C2	1.333 (2)	C15—H15	0.9300
N2—C3	1.329 (3)	C16—C17	1.352 (5)
N2—C2	1.345 (2)	C16—H16	0.9300
N3—C5	1.140 (3)	C17—C18	1.355 (4)
C1—C4	1.396 (3)	C17—H17	0.9300
C3—C4	1.399 (3)	C18—C19	1.381 (4)

C3—C20	1.501 (3)	C18—H18	0.9300
C4—C5	1.428 (3)	C19—H19	0.9300
C6—C7	1.383 (3)	C20—C21	1.511 (3)
C6—C11	1.381 (3)	C20—H20A	0.9700
C7—C8	1.372 (3)	C20—H20B	0.9700
C7—H7	0.9300	C21—C22	1.529 (3)
C8—C9	1.384 (3)	C21—H21A	0.9700
C8—H8	0.9300	C21—H21B	0.9700
C9—C10	1.390 (3)	C22—C23	1.489 (3)
C9—C12	1.499 (3)	C22—H22A	0.9700
C10—C11	1.366 (3)	C22—H22B	0.9700
C10—H10	0.9300	C23—C24	1.518 (3)
C11—H11	0.9300	C23—H23A	0.9700
C12—H12A	0.9600	C23—H23B	0.9700
C12—H12B	0.9600	C24—H24A	0.9600
C12—H12C	0.9600	C24—H24B	0.9600
C13—C14	1.491 (3)	C24—H24C	0.9600
C1—S1—C6	100.16 (9)	C19—C14—C13	122.7 (2)
C2—S2—C13	102.22 (9)	C15—C14—C13	120.3 (2)
C1—N1—C2	116.55 (16)	C16—C15—C14	120.5 (3)
C3—N2—C2	116.38 (17)	C16—C15—H15	119.7
N1—C1—C4	121.56 (17)	C14—C15—H15	119.7
N1—C1—S1	118.93 (13)	C17—C16—C15	121.0 (3)
C4—C1—S1	119.50 (15)	C17—C16—H16	119.5
N1—C2—N2	126.81 (19)	C15—C16—H16	119.5
N1—C2—S2	118.82 (14)	C16—C17—C18	119.8 (3)
N2—C2—S2	114.34 (14)	C16—C17—H17	120.1
N2—C3—C4	121.19 (18)	C18—C17—H17	120.1
N2—C3—C20	117.6 (2)	C17—C18—C19	119.8 (3)
C4—C3—C20	121.2 (2)	C17—C18—H18	120.1
C1—C4—C3	117.46 (18)	C19—C18—H18	120.1
C1—C4—C5	120.89 (19)	C14—C19—C18	121.8 (3)
C3—C4—C5	121.57 (19)	C14—C19—H19	119.1
N3—C5—C4	179.3 (3)	C18—C19—H19	119.1
C7—C6—C11	119.6 (2)	C3—C20—C21	112.55 (18)
C7—C6—S1	119.71 (16)	C3—C20—H20A	109.1
C11—C6—S1	120.65 (16)	C21—C20—H20A	109.1
C8—C7—C6	119.4 (2)	C3—C20—H20B	109.1
C8—C7—H7	120.3	C21—C20—H20B	109.1
C6—C7—H7	120.3	H20A—C20—H20B	107.8
C7—C8—C9	122.0 (2)	C20—C21—C22	111.03 (18)
C7—C8—H8	119.0	C20—C21—H21A	109.4
C9—C8—H8	119.0	C22—C21—H21A	109.4
C8—C9—C10	117.3 (2)	C20—C21—H21B	109.4
C8—C9—C12	121.7 (2)	C22—C21—H21B	109.4
C10—C9—C12	121.0 (2)	H21A—C21—H21B	108.0
C11—C10—C9	121.5 (2)	C23—C22—C21	113.75 (19)
C11—C10—H10	119.3	C23—C22—H22A	108.8

C9—C10—H10	119.3	C21—C22—H22A	108.8
C10—C11—C6	120.11 (19)	C23—C22—H22B	108.8
C10—C11—H11	119.9	C21—C22—H22B	108.8
C6—C11—H11	119.9	H22A—C22—H22B	107.7
C9—C12—H12A	109.5	C22—C23—C24	112.6 (2)
C9—C12—H12B	109.5	C22—C23—H23A	109.1
H12A—C12—H12B	109.5	C24—C23—H23A	109.1
C9—C12—H12C	109.5	C22—C23—H23B	109.1
H12A—C12—H12C	109.5	C24—C23—H23B	109.1
H12B—C12—H12C	109.5	H23A—C23—H23B	107.8
C14—C13—S2	108.96 (14)	C23—C24—H24A	109.5
C14—C13—H13A	109.9	C23—C24—H24B	109.5
S2—C13—H13A	109.9	H24A—C24—H24B	109.5
C14—C13—H13B	109.9	C23—C24—H24C	109.5
S2—C13—H13B	109.9	H24A—C24—H24C	109.5
H13A—C13—H13B	108.3	H24B—C24—H24C	109.5
C19—C14—C15	117.1 (2)		
C2—N1—C1—C4	-0.9 (2)	C6—C7—C8—C9	0.4 (3)
C2—N1—C1—S1	178.62 (12)	C7—C8—C9—C10	1.2 (3)
C6—S1—C1—N1	-10.82 (15)	C7—C8—C9—C12	-178.7 (2)
C6—S1—C1—C4	168.75 (14)	C8—C9—C10—C11	-1.7 (3)
C1—N1—C2—N2	-0.9 (3)	C12—C9—C10—C11	178.3 (2)
C1—N1—C2—S2	-179.06 (12)	C9—C10—C11—C6	0.6 (3)
C3—N2—C2—N1	1.2 (3)	C7—C6—C11—C10	1.1 (3)
C3—N2—C2—S2	179.48 (13)	S1—C6—C11—C10	-178.29 (15)
C13—S2—C2—N1	-1.69 (16)	C2—S2—C13—C14	156.66 (16)
C13—S2—C2—N2	179.92 (14)	S2—C13—C14—C19	75.3 (2)
C2—N2—C3—C4	0.2 (3)	S2—C13—C14—C15	-104.8 (2)
C2—N2—C3—C20	-179.37 (16)	C19—C14—C15—C16	0.2 (4)
N1—C1—C4—C3	2.2 (3)	C13—C14—C15—C16	-179.7 (2)
S1—C1—C4—C3	-177.32 (13)	C14—C15—C16—C17	0.1 (5)
N1—C1—C4—C5	179.03 (17)	C15—C16—C17—C18	-0.4 (5)
S1—C1—C4—C5	-0.5 (2)	C16—C17—C18—C19	0.4 (4)
N2—C3—C4—C1	-1.9 (3)	C15—C14—C19—C18	-0.2 (3)
C20—C3—C4—C1	177.73 (17)	C13—C14—C19—C18	179.6 (2)
N2—C3—C4—C5	-178.63 (18)	C17—C18—C19—C14	0.0 (4)
C20—C3—C4—C5	1.0 (3)	N2—C3—C20—C21	-85.7 (2)
C1—S1—C6—C7	-85.01 (17)	C4—C3—C20—C21	94.7 (2)
C1—S1—C6—C11	94.37 (16)	C3—C20—C21—C22	-170.5 (2)
C11—C6—C7—C8	-1.6 (3)	C20—C21—C22—C23	-173.8 (2)
S1—C6—C7—C8	177.82 (16)	C21—C22—C23—C24	179.6 (2)